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Phosphate availability and the ultimate control of new nitrogen input by nitrogen fixation in the tropical Pacific Ocean

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Abstract. Due to the low atmospheric input of phosphate into the open ocean, it is one of the key nutrients that could ultimately control primary production and carbon export into the deep ocean. The observed trend over the last 20 years has shown a decrease in the dissolved inorganic phosphate (DIP) pool in the North Pacific gyre, which has been correlated to the increase in di-nitrogen (N₂) fixation rates. Following a NW-SE transect, in the Southeast Pacific during the early austral summer (BIOSCOPE cruise), we present data on DIP, dissolved organic phosphate (DOP) and particulate phosphate (PP) pools along with DIP turnover times (T_{DIP}) and N₂ fixation rates. We observed a decrease in DIP concentration from the edges to the centre of the gyre. Nevertheless the DIP concentrations remained above 100 nmol L⁻¹ and T_{DIP} was more than 6 months in the centre of the gyre; DIP availability remained largely above the level required for phosphate limitation to occur and the absence of *Trichodesmium* spp and low nitrogen fixation rates were likely to be controlled by other factors such as temperature or iron availability. This contrasts with recent observations in the North Pacific Ocean at the ALOHA station and in the western Pacific Ocean at the same latitude (DIAPALIS cruises) where lower DIP concentrations (<20 nmol L⁻¹) and T_{DIP}<50 h were measured during the summer season in the upper layer. The South Pacific gyre can be considered a High Phosphate Low Chlorophyll (HPLC) oligotrophic area, which could potentially support high N₂ fixation rates and possibly carbon dioxide sequestration, if the primary ecophysiological controls, temperature and/or iron availability, were alleviated.

1 Introduction

New nitrogen (N) input by N₂ fixation has been recognized as a significant process influencing global oceanic productivity and the associated carbon fluxes (Karl et al., 1997; Falkowski, 1997). It is the primary process responsible for the input of N-containing compounds into the sea (Codispoti et al., 2001; Capone and Knapp, 2007) and can decouple N from other bioelement cycles which depend on ocean mixing; this decoupling has potentially important biogeochemical implications (Gruber, 2004; Karl, 2007). This input largely depends on the biomass of N₂ fixing organisms, which in turn depends on factors such as temperature (Capone et al., 1997; Staal et al., 2003; Breitbarth et al., 2007), iron (Falkowski, 1997; Kustka et al., 2002; Fu and Bell, 2003) and P availability (Karl et al., 1997; Sanudo-Wilhelmy et al., 2001; Moutin et al., 2005; Karl et al., in press). These factors are affected by human activity (increase in temperature, increase in nutrient input by rivers and changes in atmospheric Fe input, Tagliabue et al., 2007), so the input of N, via N₂ fixation, may change over time. Indeed, Karl et al. (1997) have observed an increase in diazotrophic biomass at the ALOHA station, in the Subtropical North Pacific gyre, along with a decrease in soluble reactive P (SRP which is equivalent to DIP in our study) and an apparent shift from N limitation to P limitation (Karl et al., 2001; Karl, 2007).

P availability is a crucial factor in controlling the process of N₂ fixation. It has long been considered, by “geochemists”, as the ultimate factor controlling primary production in the global ocean (Redfield et al., 1963; Tyrrell, 1999). Contrary to N, there is a negligible atmospheric reservoir of P so there is no alternative source when P runs out. Furthermore, P availability in the open ocean is probably less affected by human activity. P coming into the sea is mainly of river origin (Broecker and Peng, 1982). Riverine fluxes of

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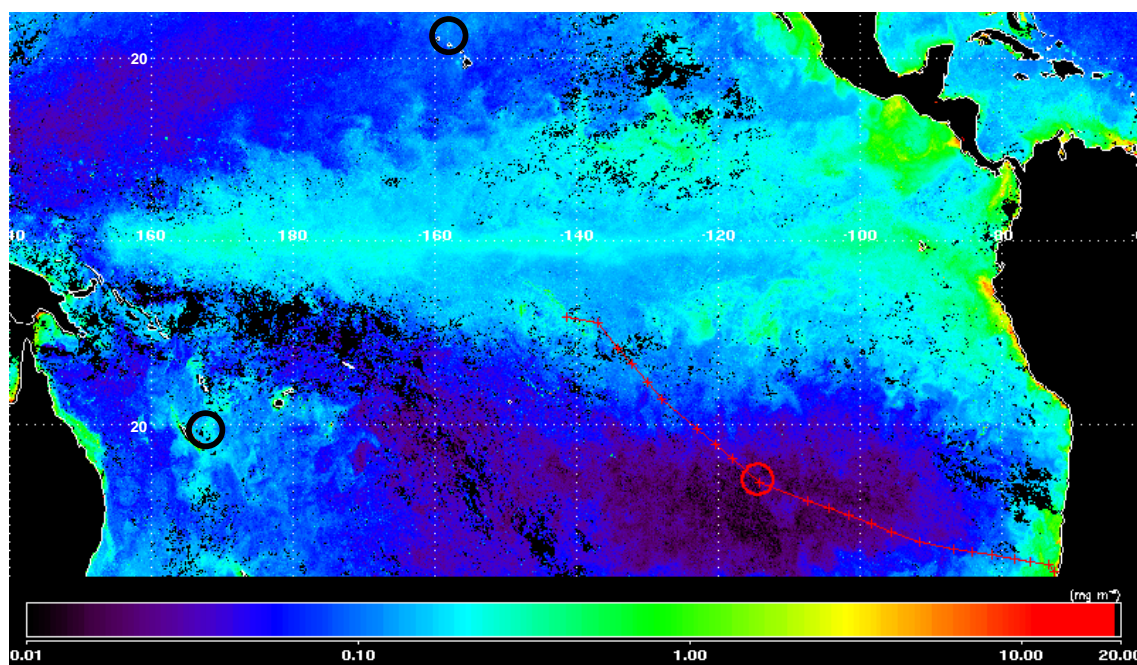


Fig. 1. Location of the stations sampled during the BIOSOPE cruise (November–December 2004) and of the three stations (red circle) where current P pools and availability are compared: a station in the centre of the South gyre (S-gyre station), ALOHA station frequently sampled since the 80's (N-gyre station) and a station sampled during the DIAPALIS cruises (2001–2002) in the SW Pacific Ocean (SW station).

P are also increasing (Meybeck, 1993; Moutin et al., 1998), however, due to its rapid consumption by biological or chemical processes in estuaries (Lucotte and Danglejan, 1988; Golterman and de Oude, 1991), P input by rivers, may have little impact on a global scale, at least over a decadal time scale. Most P is trapped in the superficial sediment (Van Den Broeck and Moutin, 2002; Paytan and McLaughlin, 2007) and may only have local impact. Thus, it is necessary to determine the current P availability in the open ocean to determine its role in the control of N₂ fixation, at the present time and in the near future.

We studied N₂ fixation, P pools and subsequent nutrient availability for planktonic species, following a NW–SE transect in the South Pacific Ocean during November–December 2004 (BIOSOPE cruise). N₂ fixation rate, and DIP, DOP and PP pools were measured together with DIP turnover times in the upper water column. Data from this cruise, particularly from the central gyre, which is one of the least studied major oceanic entities of the world's ocean, is compared with data obtained from the North Pacific gyre at the ALOHA station and a station in the South Western Pacific, where blooms of the N₂ fixing cyanobacterium *Trichodesmium*, are frequently observed (Dupouy et al., 2000). Other diazotrophs such as the recently described unicellular cyanobacteria (Zehr et al., 2001; Zehr et al., 2007) further emphasise the importance of the N₂ fixation process in the budget of new N. Nevertheless, current estimates suggest that *Trichodesmium* may be two- to threefold more abundant than

previously reported and may account for the missing sink of ~90 Tg N required to support the observed new production in the ocean (Davis and McGillicuddy, 2006; Levitan et al., 2007) and confirms the prevalent role of *Trichodesmium* previously reported “in the oligotrophic ocean” (Capone et al., 1997). We consider the current understanding of the factors controlling N₂ fixing organism biomass and the most probable temporal evolution of these factors, then discuss the predictable changes that may occur in N₂ fixation and DIP availability in the south equatorial Pacific.

2 Methods

2.1 Station locations, cruises, chronology and sample collection

The BIOSOPE cruise was carried out during November–December 2004 from the Marquesas Islands to the Chilean coast (Fig. 1). We sampled water along a 6000 km transect and the clearest ocean waters, near Easter Island (Morel et al., 2007) inside the South East Pacific gyre suggested to be the most oligotrophic and stable water body on the basis of remotely sensed ocean colour (Claustre and Maritorena, 2003). All samples were collected from a CTD rosette fitted with 20, 12-L Niskin bottles equipped with silicone rubber closures and tubing that had been carefully cleaned to avoid introducing toxic metals during sampling. Following water collection, samples were processed within 1 h.

Table 1. SST range (°C, monthly long-term mean data), and integrated (0–100 m integrals) P pools (mmol m⁻²) at the three studied stations. N-gyre (ALOHA station, <http://www.soest.hawaii.edu>) and SW station (Chenal des Loyauté station, <http://www.com.univ-mrs.fr/IRD/urcyano/identite/spdiap.htm>). *NOAA_OI_SST_V2 data provided by the NOAA/OAR/ESRL PSD, Boulder, Colorado, USA, from their Web site at <http://www.cdc.noaa.gov/>

		SW station	N-gyre station (ALOHA)	S-gyre station
		21° 30 S; 167° E	22°45 N; 158° W	26°05 S; 114° W
SST (°C) range*	Min–Max	23.1–27.3	23.4–26.4	20.7–25.1
I DIP (0–100 m)	Winter–Summer	9.65–5.40	5.07–5.06	12.45
I DOP (0–100 m)	Winter–Summer	26.45–21.35	21.66–23.69	16.28
I PP (0–100 m)	Winter–Summer	1.97–1.52	1.44–1.33	0.54

Data from the central station, inside the South Pacific gyre (S-gyre station), are compared with data obtained from the DIAPALIS cruises, at the chenel des Loyauté station (SW station) in 2002–2003, and data from the ALOHA station (N-gyre station) in the North Pacific in 2000–2001 (Table 1 and Fig. 1).

2.2 Analytical methods

2.2.1 N₂ fixation rate

One mL of ¹⁵N₂ gas (99% ¹⁵N₂ EURISOTOP) was introduced to each 0.6 L polycarbonate bottle through a Teflon-lined butyl rubber septum using a gas-tight syringe, following the protocol of Montoya et al. (1996). Following 24 h incubations, the samples were filtered under low vacuum (100 mm Hg) through precombusted (24 h at 450°C) 25 mm GF/F filters and dried at 60°C. Filters were stored in a desiccator until processed. Determination of ¹⁵N enrichments were performed using an Integra-CN PDZ EUROPA mass spectrometer. We measured a background natural abundance, determined from 8 unlabelled samples, of 0.367 ± 0.007%. Only excess enrichments higher than two times the standard deviation (0.014% for N) were considered significant. As we worked with low levels of particulate N (PN), we calibrated the spectrometer using the same conditions as for PN and the quality of the analysis was tested with standard molecules (glycine). Based on the lowest N level determined by our mass spectrometer (0.2 μmol), the detection limit for N₂ fixation was 0.12 nmol L⁻¹ d⁻¹. N₂ fixation rates (nmol L⁻¹ d⁻¹) were computed from an equation based on final PN (Dugdale and Wilkerson, 1986). A detailed description of the methodologies for ¹⁵N based measured rates including N₂ fixation rates is given in the paper of Raimbault and Garcia (2007).

2.2.2 P pools

Total P (TP) in seawater samples may be separated in three pools: the Dissolved Inorganic P (DIP) pool, the Dissolved Organic P (DOP) pool and the Particulate P (PP) pool.

The DIP pool was estimated on board using the molybdenum blue reaction (Strickland and Parsons, 1972), on 50 mL samples, using a 10 cm length-cuvette in a spectrophotometer (Cecil CE 1011), at 880 nm, following a 30 min reaction time. Concentrations were expressed in nmol L⁻¹. The lower limit of detection for DIP by this method was 20 nmol L⁻¹. Only one sample was analysed per depth with the exception of the S-gyre station where triplicate samples were analysed. The new MAGIC 25 procedure (Rimmelin and Moutin, 2005) was used on triplicate samples to determine DIP concentration (detection limit = 0.8 ± 0.5 nmol L⁻¹) and arsenate concentration at the S-gyre station. The arsenate concentration was 10.8 nmol L⁻¹ (sd = 8.6, n = 21). Because this is lower than the detection limit outlined in the Strickland and Parsons (1972) procedure for DIP measurement and is not constant (relatively high standard deviation), no arsenate correction was taken into account for DIP measurement.

The PP pool was determined by the filtration of 1-L samples through polycarbonate filters (0.2 μm; 47 mm) at very low pressure. Pressure was increased to get a drop by drop filtration but never exceeded 0.1 bar. PP was measured by standard DIP analysis at 880 nm, following high temperature persulfate wet-oxidation at 120°C and 1 bar (Pujopay and Raimbault, 1994) which converts all inorganic and organic non-reactive P compounds to DIP. Only one sample was analysed per depth with the exception of the S-gyre station where duplicate samples were analysed. Using a concentration factor of 40 (final volume = 25 mL), the detection limit was 0.5 nmol L⁻¹.

Total P (TP) was estimated from 40 mL duplicate seawater samples, using the same high-temperature persulfate wet-oxidation pre-treatment as for PP. DIP was then analysed as previously described. The measurement of DOP in seawater requires simultaneous measurements of DIP and TP. DOP was assumed to be equal to ([TP] – [DIP + PP]). The precision and accuracy of the DOP estimates decreased with increasing depth, as DIP concentrations became the dominant component in the total dissolved nutrient pools.

2.2.3 Turnover times of DIP

The DIP turnover time (T_{DIP}) is the ratio of DIP concentration to DIP uptake rate. Thingstad et al. (1993) derived the following relationship for T_{DIP} : $r(t) = 1 - e^{-t/T}$, where r is the fraction of added radioactivity absorbed, t the incubation time and T the turnover time. This equation has been rearranged in order to give a direct calculation for turnover time: $T = -t/\ln(1-R(t))$ where $R(t) = (R_f - R_b)/R_t$, R_f , the radioactivity on the filter, R_b , the radioactivity of the blank, and R_t , total tracer added. T_{DIP} was determined using ^{33}P tracer (H_3PO_4 in dilute hydrochloric acid; Amersham BF 1003; specific activity $>3000 \text{ Ci mmol}^{-1}$). The working solution of ^{33}P -DIP was prepared by diluting the tracer in Milli-Q water filtered through pre-washed $0.2 \mu\text{m}$ filters, immediately before use. 300 mL of seawater was dispensed into acid-washed Milli-Q and sample rinsed polycarbonate bottles, and incubated with $25 \mu\text{Ci}$ carrier-free $^{33}\text{PO}_4$ working solution to give a total activity of $0.08 \mu\text{Ci mL}^{-1}$. Less than 0.03 nmol L^{-1} of P was added to each sample. The bottles were then placed in an on-deck incubator and maintained at constant temperature using a continuous circulation of surface seawater, at 50, 25, 15, 7, 3 and 1% incident light. The same protocol was used for duplicate 300 mL samples where $300 \mu\text{L HgCl}_2$ (20 g L^{-1}) had been added as a control for non-biological uptake. Incubations were stopped by the addition of $600 \mu\text{L}$ of non-radioactive KH_2PO_4 (10 mmol L^{-1}). The optimal incubation time (4 to 5 h) was determined from a prior time-series experiment. Sample counts were at least 10 times greater than the blanks, less than 10% of the radioactivity in the samples was consumed, and incubations did not exceed several hours in order to minimize the increase in bacterial production caused by confinement (Van Wambeke et al., 2007). DIP turnover times were measured in 50 mL duplicate sub-samples from each bottle. Filtrations were carried out on 25 mm polycarbonate filters ($0.2 \mu\text{m}$), placed on DIP-saturated support GF/F filters, using a low-vacuum pressure <0.2 bars. Filters were not washed with filtered seawater at the end of the filtration, but pressure was briefly increased to 0.6 bars, to remove non cellular radioactivity from the filter. Filters were then placed in low-potassium 6 mL glass scintillation vials (Wheaton) with 6 mL of scintillation liquid (Ultima gold MV, Packard) and the radioactivity of the filters measured using a scintillation counter Packard Tri-Carb® 2100TR. Initial radioactivity was measured on 5 replicates of $5 \mu\text{L}$ of working solution alongside each experiment in order to verify the amount of ^{33}P added to each sample.

2.2.4 Labile DOP

We used the Strickland and Parsons (1972) procedure to measure the labile DOP within the gyre. This pool was assumed to be composed mainly of P monoesters and thus, to be easily hydrolysed by alkaline phosphatase. At each station from the

NW edge to the centre of the gyre, 50 mL triplicate surface (50% of incident light depth) samples were incubated with 1 mL of a fresh and pure solution of *Escherichia coli* alkaline phosphatase (Sigma P-4252, 0.2 U mL^{-1}) and 1 mL of Tris buffer solution 0.5 M (pH=8). Following 2 h incubation at 30°C , the DIP concentration was measured. A further set of triplicate surface samples was used to determine the initial DIP concentration. The labile DOP concentration is the difference between these two measurements. For each series of measurements, a blank sample was processed as well as a control with $1 \mu\text{mol L}^{-1}$ of a glucose-6-phosphate solution in order to confirm enzymatic efficiency.

2.2.5 Excess DIP relative to nitrate concentration

Ambient nitrate + nitrite and nitrite were measured immediately on collection by pumping the sample from the sampling polyethylene bottle directly through the Technicon AutoAnalyzer. Nitrate and nitrite concentrations in the nanomolar range (lower detection, 3 nmol L^{-1}) were obtained using the sensitive method of Raimbault et al. (1990). For submicromolar concentrations, the classical Technicon AutoAnalyzer method was used (Tréguer and LeCorre, 1975). The recently defined variable P^* (Deutsch et al., 2007) was calculated: $P^* = \text{PO}_4 - \text{NO}_3/\text{rr}$ (rr=Redfield ratio=16), PO_4 =DIP in this study. While nutrient uptake by non-N₂-fixing organisms will on average consume NO_3 and PO_4 in a proportion that conserves P^* , N₂ fixation will extract PO_4 alone, driving a water parcel towards lower P^* (Deutsch et al., 2007).

3 Results

3.1 P pools and availability in the tropical Pacific Ocean

3.1.1 Spatial distribution (SE Pacific Ocean)

DIP, DOP and PP concentrations in surface waters were highest near the Marquesas Islands and near the Chilean coast (Fig. 2a, b, c). For the 3 distinct pools, there is an obvious decrease in concentration from the edge to the centre of the gyre, reaching minimum values of 120 and 150 nmol L^{-1} for the DIP and DOP pools respectively, and less than 10 nmol L^{-1} for PP. Duhamel et al. (2007) have argued that the PP pool is mainly associated with living biomass in the centre of the gyre. Values of PP less than 10 nmol L^{-1} suggest extremely low biomass as confirmed by total chlorophyll *a* (TChl*a*) concentrations reaching 0.017 mg m^{-3} in surface waters in the centre of the gyre (Ras et al., 2007). In the upper layer from the edge to the centre of the gyre, the decreases in DIP and PP concentrations are approximately one order of magnitude, compared to a factor of 2.5 for the DOP pools. The DIP/DOP concentration ratio is close to one in the centre of the gyre. The mean labile DOP concentration (P monoesters) inside the gyre at 5 meter depth was below the detection limit of 20 nmol L^{-1} . DIP turnover

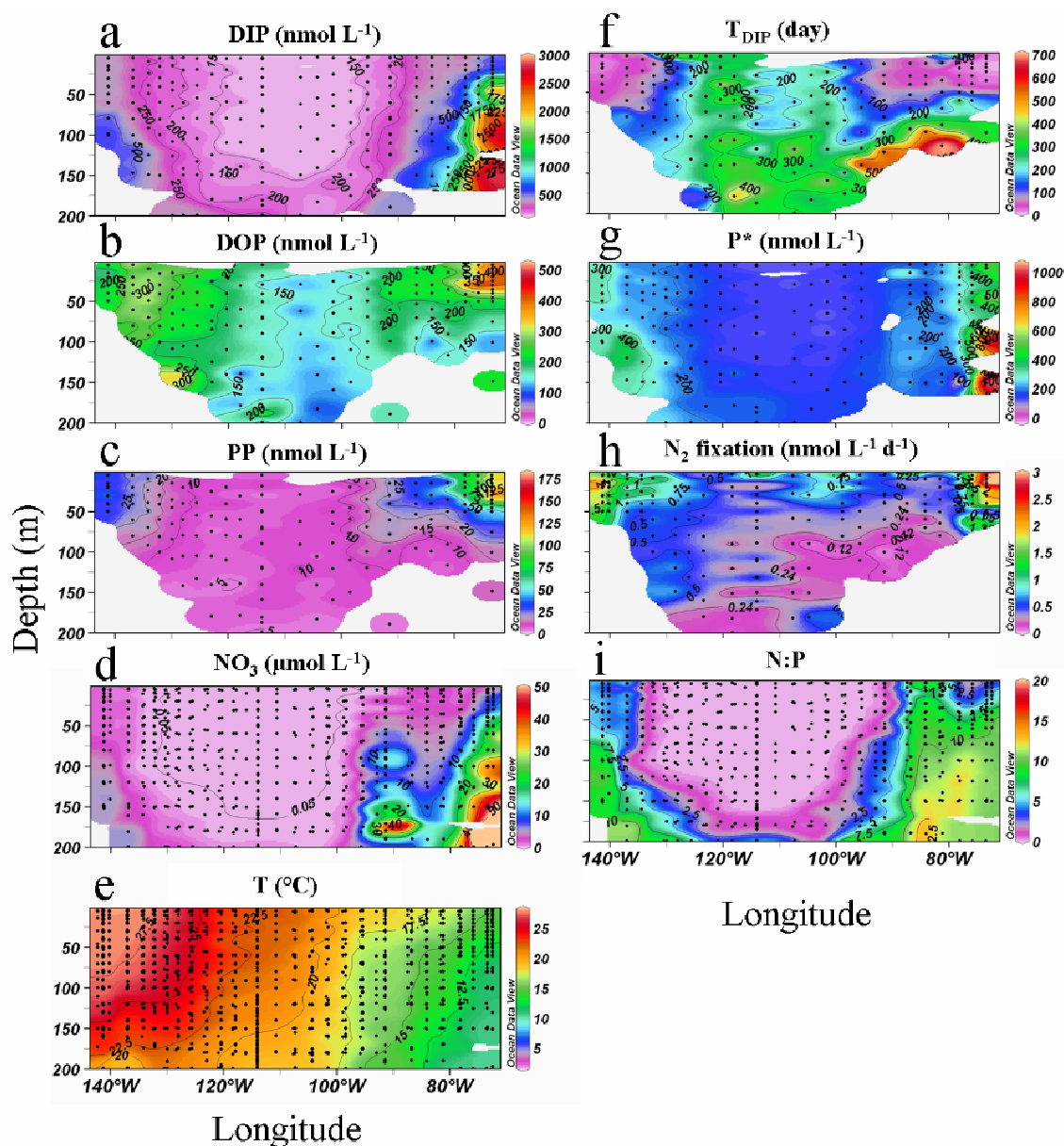


Fig. 2. (a) DIP, (b) DOP, (c) PP, (d) NO_3^- , (e) Temperature, (f) DIP turnover time, (g) P^* , (h) N_2 fixation rates and (i) NO_3^-/DIP measured during the BIOSOPE cruise (November–December 2004) in the South Pacific between the Marquesas Islands and the Chilean coast. ODV (Schlitzer, 2002) was used to generate the distribution maps.

times varied from several days at the edge, to around 200 days in the gyre (Fig. 2f). Higher P^* values were observed near the Chilean coast below 100 m depth ($\sim 1000 \text{ nmol L}^{-1}$) and close to the Marquesas Islands ($\sim 500 \text{ nmol L}^{-1}$). Values of less than 200 nmol L^{-1} were observed in the centre of the South Pacific gyre (Fig. 2g) where the nitrate concentrations became $< 50 \text{ nmol L}^{-1}$ (Fig. 2d). A significant relationship ($p < 0.001$; non parametric Pearson test) was obtained between N_2 fixation and P^* ($\text{N}_2 \text{ fixation} = 6.37 \text{ P}^* - 0.41$, $N = 136$, $r^2 = 0.49$) using all data collected during the BIOSOPE cruise.

3.1.2 Vertical depth profile inside the gyre (S-gyre station)

DIP concentrations are homogeneous having a mean value of $127.0 \text{ nmol L}^{-1}$ ($\text{sd} = 7.1$, $n = 42$) in the surface waters (0–150 m) and increasing with depth below 150 m (Fig. 3f) at the S-gyre station. The DIP concentration versus depth gradient between 150 and 250 m is $3.5 \mu\text{mol m}^{-4}$. DIP concentrations in surface waters are above the detection limits of the standard method (Strickland and Parsons, 1972). No significant difference was observed with the high sensitivity MAGIC 25 method (Rimmelin and Moutin, 2005) thus enabling us to

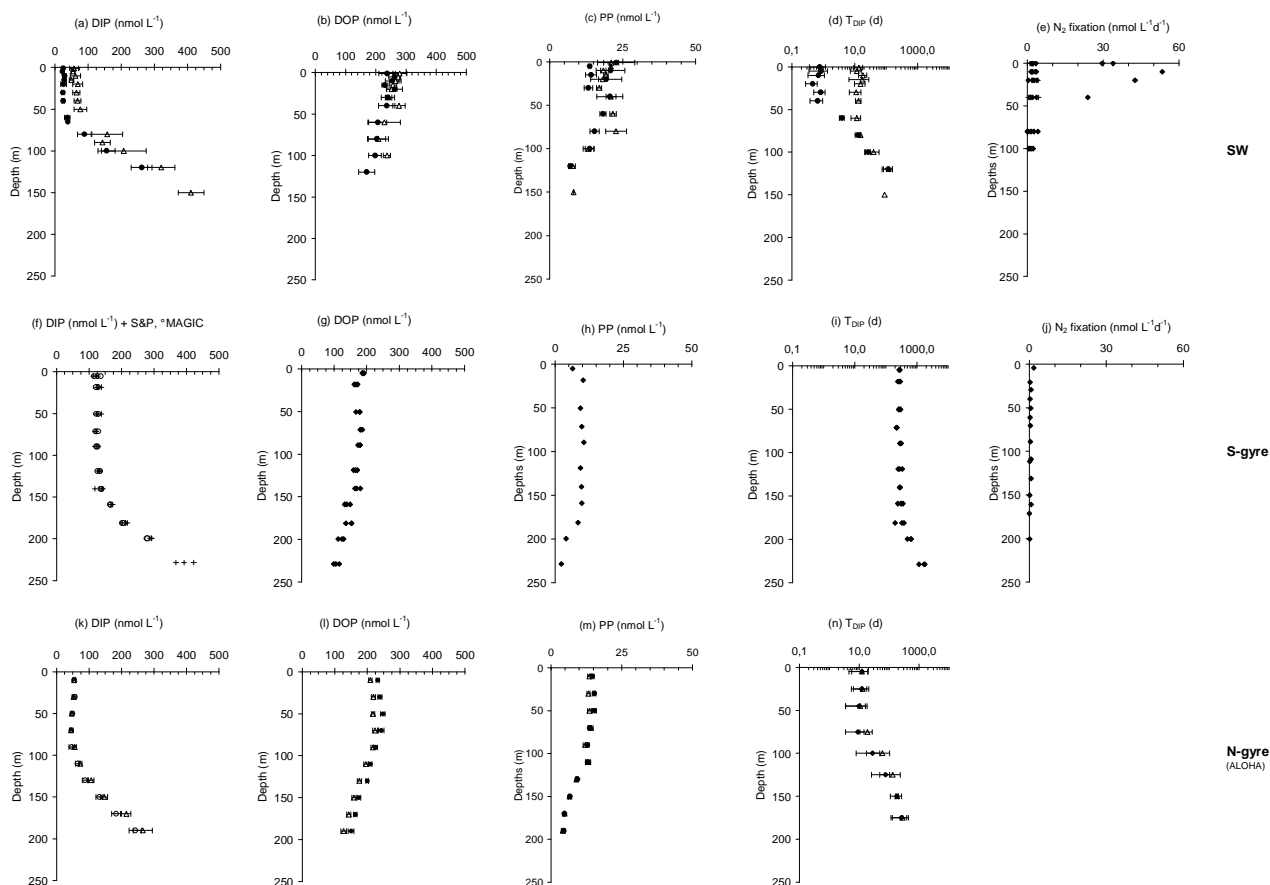


Fig. 3. Current Dissolved Inorganic P (DIP), Dissolved Organic P (DOP), Particulate P (PP) concentrations, DIP turnover time and N₂ fixation rates vs depth at the three stations studied in the South, N-gyre (ALOHA station, http://www.soest.hawaii.edu/HOT_WOCE/ftp.html) and SW station (Chenal des Loyauté station, <http://www.com.univ-mrs.fr/IRD/urcyano/identite/spdiap.htm>). Summer and Winter (Δ) concentrations are indicated when available.

validate our detailed protocol for a 25 times MAGIC pre-concentration procedure. DIP turnover times are homogeneous with a mean value of 273 days (sd=29, n=21) in surface waters (Fig. 3i).

DOP concentrations are highest in near surface waters with a mean value of 175.5 nmol L⁻¹ (sd=9.2, n=21) and decrease with increasing water depth (Fig. 3g). DOP dominates in the surface water, accounting for 58% of the total dissolved P pool. DIP becomes the dominant component of the total dissolved pool below 150 m. The mean DOP concentration from duplicate deep water samples (600–2000 m) is 61.0 nmol L⁻¹ (sd=16.5, n=8). PP concentrations follow the same pattern as DOP concentrations with a mean value of 9.3 nmol L⁻¹ (sd=1.5, n=6) in surface waters decreasing as water depth increased (Fig. 3h). A maximum in PP was not observed at the depth of the deep chlorophyll maximum (180 m). Values below 200 m were similar to the mean concentration from duplicate deep water samples (600–2000 m) of 1.0 nmol L⁻¹ (sd=0.3, n=8).

3.1.3 Comparison with depth profiles from the SW and N-gyre Pacific stations

In the upper layer, the maximum values of DIP and for T_{DIP} are recorded at the S-gyre station. Depths trends similar to those at the S-gyre station, are observed for DIP, DOP and PP at the SW (Fig. 3a, b, c) and N-gyre (Fig. 3k, l, m) stations; the only exception is the sharp decrease in DIP concentration, observed at the SW station (Fig. 3a), during the summer period. The average decrease in DIP concentration between austral winter and austral summer in the upper 40 m of the water column was 35 nmol L⁻¹ at the SW station. This difference corresponds to a minimum value as DIP concentration was generally below the 20 nmol L⁻¹ detection limit of the classical Strickland and Parsons, 1972 method. No MAGIC measurements are available but indirect estimations of the DIP concentrations from T_{DIP} measurements (Van Den Broeck et al., 2004) indicated sub-nanomolar concentrations during the summer period. There was no corresponding increase in either the PP pool or the DOP pool, although large

differences in the concentration of the DOP pool were observed during both the winter and summer periods. DIP turnover times (Fig. 3d) vary from approximately 10 days in winter to a few hours in summer in near surface waters.

No clear seasonal variations were observed in the DIP pool at the N-gyre station (ALOHA, Fig. 3k). DIP concentrations were lower (by a factor of 2–2.5) than concentrations measured in the mixed layer of the S-gyre station. DOP concentrations were slightly above the DIP concentrations measured at the S-gyre station and very close to those measured at the SW station. The PP concentrations at the N-gyre station were higher (by a factor of 1.5) than those in the more oligotrophic S-gyre station and close to or lower than those at the SW-station. DIP turnover times were around 10 days near the sea surface at the N-gyre station, close to those observed during the winter season at the SW station and no marked variations between the summer and winter seasons are observed.

3.2 Current distribution of N₂ fixation

The N₂ fixation in surface waters during the BIOSOPE cruise yielded maximum values, around 4 nmol L⁻¹ d⁻¹, near the Chilean coast, intermediate values near the Marquesas Islands (1.2–2.8 nmol L⁻¹ d⁻¹) and low values, just above the detection limit of the method employed (0.12 nmol L⁻¹ d⁻¹), inside the gyre (Fig. 2h). Integrated depth profiles gave a value of $\sim 142 \mu\text{mol N m}^{-2} \text{ d}^{-1}$ near the Chilean coast (1 in situ depth profile between 0 and 40 m) and 48 and 135 $\mu\text{mol N m}^{-2} \text{ d}^{-1}$ at the S-gyre station (2 in situ depth profiles between 0 and 200 m taken on two days). A doubling of the mean irradiance intensity and similar variations of ammonium and nitrate uptake rates were also observed between the two different days. Maximum N₂ fixation rates were found in the upwelling area where N:P ratio close to 10 (Fig. 2i) and ample Fe concentrations (1.2–3.9 nM) are found (Blain et al., 2007), and temperatures did not exceed 16° C (Fig. 2e). The N₂ fixation rates are maximal near the surface within the gyre and decrease with depth indicating that light could play a role in controlling this process. At present we do not know the species responsible for these fluxes. *Trichodesmium*, a large filamentous N₂ fixing microorganism that often forms large easily identified colonies, was not observed. Bonnet et al. (2007) found the presence of extremely low numbers of Group A cyanobacterial phylogenotypes (see their Table 2).

The N₂ fixation rates at the SW station (DIAPALIS cruises) were highly variable within seasons and ranged between 151–703 $\mu\text{mol N m}^{-2} \text{ d}^{-1}$ (Garcia et al., 2007). The maximum values correspond to a *Trichodesmium* specific bloom that occurred during the summer period (Fig. 3e) when temperatures were above 26° C. At the N-gyre station ALOHA, higher N₂ fixation rates than those measured in the South Pacific gyre have already been reported and these appear to be associated with *Trichodesmium* blooms (Dore et

al., 2002). At least two independent microbial assemblages and ecosystem processes contribute to N₂ fixation in the NP gyre, namely the “background state” wherein a relatively low but relatively constant rate supported by pico and nano-diazotrophs, and the aperiodic “bloom state” wherein large filamentous, colonial and aggregate forming diazotrophs (*Trichodesmium* and/or endosymbiont-containing diatoms) dominate the new N cycle (Karl et al., in press). It is estimated that the summer bloom is responsible for up to 38% of the annual N₂ fixation at ALOHA station (J. Dore, personal communication). Based on a variety of independent estimates including nitrogenase activity by acetylene reduction method, *Trichodesmium* abundance, N:P mass balance, and ¹⁵N isotope balance, an annual N₂ fixation rate of 31–51 mmol N m⁻² year⁻¹ was estimated for the Pacific Ocean near Hawaii (Karl et al., 1997). N₂ fixation at Station ALOHA would equate to 3–4% of the total N demand for the microorganisms that inhabit that ecosystem. However, when compared to estimates of new production or to N exports by sinking particles and migrant zooplankton, N₂ fixation appears to be a significant (40–60%) source of new N (Karl et al., in press) since export from oligotrophic ecosystems is low. When compared to estimates of new production with ¹⁵N, N₂ fixation accounted for up to 50% in the South Pacific gyre (Raimbault and Garcia, 2007). The N₂ fixation rates measured inside the SP gyre are close to the averaged $0.25 \pm 0.05 \text{ nmol N L}^{-1} \text{ d}^{-1}$ measured in oligotrophic waters of the NP Ocean where water temperatures are typically below 20–25° C (Needoba et al., 2007). Integrated rates in the SP gyre are similar to those reported by Falcon et al. (2004) for the tropical NA in summer (62 $\mu\text{mol N m}^{-2} \text{ d}^{-1}$, SE=21, n=30) and February 2001 (167 $\mu\text{mol N m}^{-2} \text{ d}^{-1}$, SE=49, n=30) and in the subtropical NP (84 $\pm 50 \mu\text{mol N m}^{-2} \text{ d}^{-1}$).

4 Discussion

4.1 P availability in the contemporary ocean

4.1.1 Availability of DIP

DIP is directly available for all microbes and its concentration can be considered as the first criterion to evaluate P availability. DIP is not completely exhausted from the upper water at the N-gyre station (ALOHA) or in the South Pacific gyre, but DIP concentrations reach <20 nM concentration during the summer season at the SW station. As these DIP concentrations at the SW station are close to or even below detection limits of classical chemical analyses (20 nM), the study of DIP turnover times can provide additional information on nutrient availability: DIP turnover time is the most broadly-applicable measurement of DIP availability because it has the potential to identify variations in P availability even when DIP concentrations become analytically undetectable (Moutin et al., 2002; Moutin et al., 2005).

Table 2. DIP turnover time (mean and standard deviation in hours) in upper ocean surface waters (0–20 m or at the depth of 50% of incident light*), D5S1N5 is for 5 depths (D5), 1 station sampled (S1), 5 samples (N5), as an example. **=median value. See also (Tanaka et al., 2006) for others T_{DIP} measurements in the Mediterranean Sea.

Winter	Spring	Summer	Fall	Area	Year of sampling	References
396 (370)	3.9 (2.4)	1.6 (0.9)		Mediterranean Sea		
D1S9N9	D1S17N17	D1S20N20		Lion gulf	1997–1998	Moutin et al. (2000)
			10.5 and 7.6	Alboran	1999	Moutin et al. (2002)
			D1S2N2			
			1.7 (1.5)	Western basin	1999	Moutin et al. (2002)
			D1S5N5			
			1.2 (0.5)	Eastern basin (Ionian)	1999	Moutin et al. (2002)
			D1S4N4			
2.9 and 6.7				Eastern basin (Ionian)	1996	Zohary and Robarts (1998)
D1S2N2						
4.1 (1.6)				Eastern basin (Levantin)	1996	Zohary and Robarts (1998)
D1S5N5						
11		5		Atlantic Ocean		
D1S1N1		D1S1N1		Sargasso Sea (BATS)	1992–1993	Cotner et al. (1997)
230 (38)	1.3 (0.4)		217	Gascogne gulf		Labry et al. (2002)
D1S3N3	D1S4N4		D1S1N1			
			82 (25)	Marocco upwelling	1999	Moutin, unpublished
			D3S1N3			
1182 (382)	152 (55)		168 (110)	North Eastern	2000–2001	Moutin, unpublished
D2S3N6	D2S4N8		D2S4N8	(between Azore islands and Spain)		
				Pacific Ocean		
504 (175)	107 (37)	10.8 (2.4)	4.0 (0.3) and 68 (2)	South Western	2002–2003	Van den Broeck et al. (2004)
D5S1N5	D5S1N5	D3S2N6	D5S1N5			
672**	456**	744**		Northern central	1973–1974	Perry and Eppley (1981)
S5	S6	S4		(within a degree of 30° N, 155° W)		
	504 (399)	144	72	Northern central gyre	1996–1997	Björkman et al. (2000)
	D1S1N3	D1S1N1	D1S1N1	(ALOHA)		
	216 and 48			Northern central gyre	1996–1997	Björkman et al. (2000)
	D1S2N2			(Climax)		
514 and 500	291 (36)	93 (35)	405 (31)	Northern central gyre	2000–2001	Björkman et al. (2003)
D2S1N2	D2S3N6	D2S2N4	D2S2N4	(ALOHA)		
149	120	221		Northern central gyre	2002–2004	Van Mooy, unpublished
D1S1	D1S2	D1S1		(ALOHA)		
	936 (451)			South equatorial	2004	This study*
	D1S3N6			(Marqueous Island)		
	2086 (1202)			Western transition area	2004	This study*
	D1S5N10					
	5589 (1472)			Transect inside the southern gyre	2004	This study*
	D1S8N16					
	1964 (1003)			Eastern transition area	2004	This study*
	D1S5N10					
	464 (305)			Chilean upwelling	2004	This study*
	D1S3N6					

DIP turnover times represent the ratio between DIP concentration and DIP uptake by the microbial assemblage. Despite the DIP concentrations being much lower in the centre of the South gyre than those found near the Marquesas Islands and Chilean coast, the larger T_{DIP} indicates that DIP availability, compared to the planktonic species requirement within the gyre, is greater than in the upwelling region. The turnover time is also the time it would take for all the ambient DIP to be taken up assuming no additional

input (Ammerman et al., 2003). Without any additional external sources or input by regeneration, it would take 200 days to exhaust all available DIP in the gyre, while it may only take 10 days in the Chilean upwelling and at the N-gyre station (ALOHA). Nonetheless, T_{DIP} in the Southern and Northern gyres suggest a P sufficiency that is contrary to what has been observed in the very P-depleted Mediterranean Sea, Sargasso Sea and South Western Pacific during the summer season (Table 2) where T_{DIP} of less than 10 h

were observed (Table 2). Nevertheless, short DIP turnover times of around 2 days (Bjorkman et al., 2000, Table 2) and a long term decrease in DIP availability (Karl et al., 1997; Karl et al., 2001) have also been reported at the ALOHA station. The comparison between DIP turnover times obtained by Perry and Eppley (1981) in the central Pacific gyre in the 70's, and current estimations (Table 2) confirm the decrease in DIP availability, particularly during the summer season. The large differences in T_{DIP} measurements observed in oligotrophic areas (Table 2) enable us to distinguish the High P Low Chlorophyll (HPLC) oligotrophic areas (SP gyre) and Low P Low Chlorophyll (LPLC) oligotrophic areas (Sargasso Sea, Mediterranean Sea). The ALOHA station in the NP gyre appears to be an intermediary or transient state of DIP availability.

4.1.2 Availability of DIP and N₂ fixation

Low DIP availability may control the growth and biomass of N₂ fixing organisms. A critical DIP turnover time of 50 h, giving favourable and unfavourable growth conditions for *Trichodesmium*, was determined during the DIAPALIS survey in the SW Pacific ocean (Moutin et al., 2005). It was demonstrated that the sharp decrease in DIP availability in the early summer season could explain most of the numerous and periodic sea surface accumulations of *Trichodesmium* observed, and more importantly, the prevalent role of DIP availability in the control of N input by N₂ fixation in this area (Moutin et al., 2005). The DIP turnover times of around 10 days measured at the N-gyre station (ALOHA, Fig. 3n) suggests a higher P deficiency than in the South Pacific gyre (T_{DIP} around 200 days and DIP concentrations $>100 \text{ nmol l}^{-1}$), but this is not strong enough to provoke *Trichodesmium* decay (obtained for $T_{\text{DIP}} <50 \text{ h}$ equivalent to DIP concentration $<10 \text{ nmol l}^{-1}$ at the SW station). Thus, another factor must prevent the complete exhaustion of DIP and may currently control N₂ fixation by *Trichodesmium* at the N-gyre station (ALOHA). Nevertheless, episodic *Trichodesmium* blooms followed by severe DIP depletion have already been observed at the N-gyre station (Letelier, ASLO meeting 2006).

Due to their smaller size, small diazotrophs are probably more efficient in taking up DIP from the surrounding water and as a consequence, are probably less affected by DIP depletion. It is likely that the current high DIP availability in the SP gyre may not control N₂ fixation, at least during the studied period (November–December). Strong seasonal variations in DIP availability are not expected as this is an area where the lowest sea surface chl *a* concentrations in the world's oceans are determined from space throughout the year (Claustre and Maritorena, 2003).

4.1.3 Availability of DOP

For the most part DOP is not directly available to living organisms as it cannot be taken into the cell in this form (Cembella et al., 1984; Bjorkman and Karl, 2003; Paytan and McLaughlin, 2007). Organic P must first be converted to DIP. The labile fraction of the DOP pool inside the South gyre was $<20 \text{ nmol l}^{-1}$ suggesting that only a small fraction of the DOP pool was available (through alkaline phosphatase activity). At the SW station during the summer period, DIP concentrations reached $<20 \text{ nmol l}^{-1}$. However, no significant change was observed in the DOP concentrations (Fig. 3b), which suggests that most of the DOP pool in the upper surface is not readily available, even after several months of severe DIP depletion. Moreover, despite the variations in DIP concentrations and turnover times at all three stations, the DOP concentration was around 200 nmol l^{-1} at all three suggesting that the DOP reservoir is not a particularly dynamic reservoir of P in upper ocean waters. The South Pacific gyre appears to be an ocean as close as can be expected (Lewis, personal communication). Thus, the DOP pool in the S-gyre station may mainly be formed by an accumulation of older organic molecules with low bioavailability.

Even though most of the DOP pool is not readily available, it may play a determinant role in phytoplankton growth. It is probable that the turnover of the available DOP pool is rapid. It has been estimated that DOP utilization can be of the same order of magnitude as DIP utilization in the upper water column at ALOHA (Bjorkman and Karl, 2003). Recent evidence also indicates that certain compounds in the DOP pool, particularly the phosphonates, which may account for 25% of the high molecular weight DOP pool (Clark et al., 1998; Kolowitz et al., 2001), may be used (Dyhrman et al., 2006), but we know little about its global significance and there is a major scope for future research in the field of P availability.

4.1.4 Relative availability of DIP pool vs availability of DIN (DIN: Dissolved Inorganic Nitrogen which is equivalent to nitrate in our study)

P^* is a convenient estimate of the excess in DIP relative to DIN, when it is assumed that these nutrients are utilised following Redfield proportions (N:P=16). P^* variations in the South Pacific are close to those expected by Deutsch et al. (2007). They argue that denitrification in the oxygen minimum zones (OMZs) generates DIP-enriched and DIN-deficient waters. These waters subsequently undergo a disproportionate loss of DIP as they are upwelled and transported into the adjacent gyres. The eastern Pacific Ocean near the Chilean coast is one of the three major areas of denitrification in the world's oceans (Codispoti and Richards, 1976). Thus, it may explain the high P^* observed (Fig. 2g). North of 14° S, the waters are under the influence of the equatorial regime (Claustre et al. this volume). Following

general circulation models, water near the Marquesas Islands (Fig. 2g) are influenced by waters upwelled near the equator. This may also explain the high P* values observed in this area. The decrease in P* toward the centre of the gyre corresponds to the trend observed by Deutsch et al. (2007): downstream of the OMZs, surface waters that initially carry a surplus of phosphorus (because of subsurface denitrification) lose this excess gradually through N₂ fixation. This may be attributed to N₂ fixation restoring the system to a “Redfieldian” balance as already suggested by Redfield (1934) and Capone and Knapp (2007).

The model by Deutsch et al. (2007) for N₂ fixation rates along the studied transect, with minimum values found on the edge and maximum values in the centre of the gyre, is contrary to our observations. They calculated a basin-wide N₂ fixation rate for the Pacific of 48 mmol N m⁻² yr⁻¹ similar to previous estimates ~50 mmol N m⁻² yr⁻¹, Redfield et al. (1963), with maximum values of up to 120 mmol N m⁻² yr⁻¹ inside the South Pacific gyre. Assuming no seasonal variations (i.e. 120000/365 ~ 330 μmol N m⁻² d⁻¹) and a maximum depth of 200 m available for N₂ fixation inside the gyre, this rate is equivalent to 1.7 nmol N L⁻¹ d⁻¹, a value at least 2.5 times greater than the mean rate measured in this study (0.24 and 0.67 nmol N L⁻¹ d⁻¹). The annual rate calculated near the Chilean coast by Deutsch et al. (2007) is under 20 mmol N m⁻² yr⁻¹. Assuming no seasonal variations (i.e. 20000/365 ~ 55 μmol N m⁻² d⁻¹) and a maximum depth of ~40 m available for N₂ fixation in the upwelling, this rate is equivalent to ~1 nmol N L⁻¹ d⁻¹, a value that is around 4 times lower than the measured value. Thus, even if the calculated (Deutsch et al., 2007) and observed (Fig. 2g) P* distributions are very similar, the N₂ fixation rates measured in the South Pacific gyre are not consistent with the predicted rates. Some of the discrepancy between observed N₂ fixation rates and model outputs is certainly due to uncertainties in ocean circulation and care should be taken, also, that the trend suggested by Deutsch et al. (2007) may not directly be reflected in a snap shot observation. The strong relationship between P* and N₂ fixation rates observed (Fig. 2g and h) may further strengthen the idea that biological N₂ fixation and marine N removal are tightly coupled (Deutsch et al., 2007): N₂ fixation begins, and is highest, when upwelled waters reach the surface near the Chilean coast. The waters do not need to be transported to the warmer adjacent gyre to lose most of their excess P. N₂ fixation in the surface waters increases the deficit in DIP relative to DIN (decrease in P*). However, N₂ fixation is not the only process driving the decrease in P* in the water. The export of material with a N:P ratio lower than the Redfield ratio will give the same pattern. It is not known whether the greater depth of the synthesis of material that is vertically exported outside the upwelling zone produces exported material with a higher P content.

Deutsch et al. (2007) argue that the ratio of N to P in seawater may be the central factor regulating N₂ fixation and

that Fe rich dust may not exert as much influence on marine N₂ fixation (Capone and Knapp, 2007) as is currently assumed (Berman-Frank et al., 2001). The close link between denitrification and N₂ fixation is convincing but another simple explanation for the higher N₂ fixation rates near the Chilean coast could be that there is less competition for nutrients so enabling N₂ fixing organisms to thrive independently from the DIN:DIP ratio.

4.2 N₂ fixation distribution and controls other than P availability

The remaining relatively high DIP concentration in the centre of the SP gyre suggests a control other than DIP availability for N₂ fixation. It is probably related to temperature or Fe availability. Indeed, the Fe requirement of N₂ fixers must be provided by the upper photic water column and the temperature must be adequate to enable growth or N₂ fixation.

4.2.1 Temperature

There are numerous studies showing a correlation between *Trichodesmium* abundance and temperature (Capone et al., 1997; Lugomela et al., 2002; Chen et al., 2003; Moutin et al., 2005). The relationship between *Trichodesmium* distribution and sea surface temperature (SST) is so “commonly accepted” that the observed temperature distribution range (20–30° C) is used to constrain N₂ fixation in oceanic biogeochemical circulation models (OCBM) (Fennel et al., 2002; Hood et al., 2004). Temperatures above 26° C are necessary for *Trichodesmium* bloom development (Carpenter et al., 2004). The correlation between water temperature and *Trichodesmium* abundance is generally attributed to oceanographic features associated with warm waters, such as a shallow mixed layer, high light regimes and oligotrophic nutrient conditions, rather than to a direct physiological response to the temperature itself (Hood et al., 2004). Nevertheless, it was recently demonstrated that the *Trichodesmium* strain IMS-101 are adapted to optimum growth temperatures of between 24 and 30° C tolerating temperatures from 20 to 34° C (Breitbarth et al., 2007). Thus, as suggested by Capone et al. (1997), seawater temperature sets a physiological constraint on the geographical distribution of *Trichodesmium* (Breitbarth et al., 2007). Most of the numerous sea surface accumulations of *Trichodesmium* observed in the South western Pacific Ocean near New Caledonia occurred during the spring and early summer periods when temperatures were above 25° C. Monthly, mean SST's reached a maximum of 25.1° C during summer at the S-gyre station reaching 27.3 and 26.4° C at the SW and N-gyre Pacific stations, respectively. SST's were less than 25° C over a large part of the South Pacific gyre during the BIOSOPE cruise (Fig. 2e) and may explain the current absence of *Trichodesmium* in the SP gyre.

Other N₂ fixing organisms may not be affected by temperature. Indeed, N₂ fixers can be encountered at temperatures close to freezing (Zielke et al., 2002; Pandey et al., 2004). Then, another control for these species is needed to explain the low N₂ fixation rates inside the gyre, or at least to explain why DIP is not completely exhausted from the upper layer.

4.2.2 Fe availability

Numerous experimental studies attribute a dominant role of Fe availability in the control of diazotroph growth (Paerl, 1994; Falkowski, 1997; Kustka et al., 2002; Mills et al., 2004). A high Fe requirement of the enzyme nitrogenase is believed to prevent N₂ fixing organisms from alleviating widespread N limitation (Falkowski, 1997). Atmospheric input of dust, which is highly enriched in Fe relative to other nutrients, plays a significant role in the distribution of *Trichodesmium* (Orcutt et al., 2001) and on the decoupling between N-P-Fe-Si biogeochemical cycles in the ocean (Karl, 2002). High rates of N₂ fixation in the North Atlantic were considered to be the result of the unusually high growth rates of N₂ fixers, stimulated by the high Fe availability (Wu et al., 2000; Deutsch et al., 2007). Contrary to the oceanic gyres situated in the Northern Hemisphere, Fe-rich dust deposition is extremely low in the South Pacific gyre (Wagener et al., 2008) and dissolved Fe concentrations of $0.13 \pm 0.03 \text{ nmol l}^{-1}$ inside the upper water (0–80 m) of the gyre (Blain et al., 2007) may prevent the development of N₂ fixing organisms. Nevertheless, enrichment experiments showed that primary production was N-limited at the S-gyre station and no N₂ fixation was measured following dust, Fe and/or P additions (Bonnet et al., 2007). The lack of response could be related to sampling at a depth of 30 m depth using a peristaltic pump, a depth where N₂ fixation rates were extremely low and close to the detection limit of the method.

4.3 P availability and the ultimate control of N₂ fixation in the tropical Pacific ocean

Following the most probable temporal changes (over a few decades) for factors controlling N₂ fixation, several scenarios have been considered for N input via N₂ fixation and the ultimate P availability for the South Pacific Ocean:

4.3.1 Increase in temperature

The expected increase in temperature will increase stratification but is not likely to modify the upwelling intensity near the Chilean coast and thus, changes in N₂ fixation will probably occur elsewhere. If the presence of *Trichodesmium* spp. is controlled by seawater temperature, any increase in temperature will increase its distribution. Breitbarth et al. (2007) predicted an 11% areal increase of *Trichodesmium*'s potential geographic distribution following a modelled SST increase of up to 3° C by 2090, but a simultaneous decrease in the area characterized by optimum growth. High SSTs are predicted

for the western Pacific which is a characteristic province for present-day *Trichodesmium* abundance (LaRoche and Breitbarth, 2005). In this area where *Trichodesmium* biomass is already controlled by P availability, no change in N₂ fixation with increasing temperature is expected. Nevertheless, the increasing widespread high sea surface temperature is expected to increase the distribution area of *Trichodesmium* driving the system towards the ultimate P control over a larger area. An increasing trend in N₂ fixation associated with a decreasing trend in P availability may therefore be expected in the South western ocean.

4.3.2 Increase in Fe dust deposition

The dominant external input of Fe to the surface of the open ocean is the transport of aeolian dust, from the great deserts of the world (Jickells et al., 2005) and this appears to be very sensitive to climate changes (Mahowald et al., 2006). The effect of dust/Fe in subtropical gyres is a reduction in Fe limitation on N₂ fixation so increasing primary production (Falkowski et al., 1998). A higher Fe supply to the sea surface has been hypothesized to favour N₂ fixation in the continentally influenced Atlantic Ocean (Wu et al., 2000) as well as in the South Western Pacific Ocean (Van Den Broeck et al., 2004) and probably the Mediterranean Sea (Bonnet and Guieu, 2006). The increasing atmospheric input of Fe in the Northern Hemisphere has been believed to cause the decreasing trend of P availability at the N-gyre station ALOHA (Karl et al., 2001). A similar trend may be expected for the South Pacific.

4.3.3 An extension of the geographical area of denitrification

Deutsch et al. (2007) suggest that N₂ fixation is closely coupled to the generation of N-deficient waters in areas of denitrification (timescale of year to decades) and are mainly dependent on the N:P ratios of the water upwelled near the coast. As denitrification occurs mainly in suboxic zones with $[\text{O}_2] < 5 \mu \text{ mol l}^{-1}$ (Codispoti et al., 2001), any change in the spread of these areas will have significant impact on the N₂ fixation rate inside the gyres. The most probable future change is an extension of the geographical areas of denitrification. This will generate a higher volume of DIN deficient waters and consequently, according to Deutsch et al. (2007) a higher N₂ fixation rate in the South Pacific and a strengthening control of the carbon cycle by P availability.

5 Conclusions

We found that P availability was largely above the level required for P limitation on N₂ fixation in the South Pacific gyre. This is in contrast to recent observations in the SW Pacific Ocean over the same latitude. We suggest a geographical trend for limitation on N₂ fixation, from P limitation in

the West to temperature and/or Fe limitation in the central and South East Pacific Ocean. The South Pacific gyre can be considered a High P Low Chlorophyll (HPLC) oligotrophic area, which could potentially support high N₂ fixation rates if the primary control temperature and/or Fe availability were alleviated. A decrease in P availability due to an increasing input of N by N₂ fixation is the most probable decadal trend to occur following climate change.

As already mentioned by Deutsch et al. (2007), environmental controls on N₂ fixation and thus its probable response to past and future climate change would be much clearer if we knew the geographic distribution of this process in the ocean. The poor correlation between their model output and our observations for N₂ fixation in the South Pacific area emphasizes the importance of studying the geographical distribution of N₂ fixation as well as the main factors controlling this process. Increased understanding of the factors controlling the growth and biomass of N₂ fixing organisms is needed in the field in order to constrain the models.

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